ADDITIONS AND CORRECTIONS

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Jennie Moriarty, Howard Sidebottom,* John Wenger,* Abdelwahid Mellouki, and Georges Le Bras: Kinetic Studies on the Reactions of Hydroxyl Radicals with Cyclic Ethers and Aliphatic Diethers

Two errors and one omission have been found in the original article.¹

Page 1501: In Table 1, the reference compound used for 1,3dioxepane was in fact (CH₃)₂CHOCH₂CH₂OH and not (CH₃)₂-CH(OH)CH₂OCH₃. The rate coefficient for reaction of OH with (CH₃)₂CHOCH₂CH₂OH (k_5) was 21.0 × 10⁻¹² cm³ molecule⁻¹ s⁻¹, as reported by Porter et al.² The reported value of k_4/k_5 for 1,3-dioxepane, 0.58 ± 0.04, is correct. The correct entry for 1,3-dioxepane in Table 1 is the following:

ether	reference compound	k_4/k_5 ^a
1,3-dioxepane	(CH ₃) ₂ CHOCH ₂ CH ₂ OH	0.58 ± 0.04

^a Errors are twice the standard deviation and represent precision only.

Page 1501: In Table 2, the rate coefficients quoted for 1,3dioxepane are incorrect and should be $(12.2 \pm 0.9) \times 10^{-12}$ and $(13.9 \pm 0.4) \times 10^{-12}$ cm³ molecule⁻¹ s⁻¹ for relative rate and PLP-LIF measurements, respectively. The rate coefficients obtained for oxepane were inadvertently omitted from Table 2 and are $(18.4 \pm 1.4) \times 10^{-12}$ and $(18.7 \pm 0.2) \times 10^{-12}$ cm³ molecule⁻¹ s⁻¹ for relative rate and PLP-LIF measurements, respectively. The correct entries for oxepane and 1,3-dioxepane in Table 2 are the following:

ether	<i>k</i> ^a	technique ^b	reference
oxepane	18.4 ± 1.4	RR	this work
	18.7 ± 0.2	PLP-LIF	this work
1,3-dioxepane	12.2 ± 0.9	RR	this work
	13.9 ± 0.4	PLP-LIF	this work

^{*a*} Units are 10^{-12} cm³ molecule⁻¹ s⁻¹. Errors are twice the standard deviation and represent precision only. ^{*b*} RR, relative rate; PLP–LIF, pulsed laser photolysis–laser induced fluorescence.

These changes only affect the data reported in Tables 1 and 2 on page 1501 and do not alter the discussion and interpretation of the results.

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References and Notes

(1) Moriarty, J.; Sidebottom, H.; Wenger, J.; Mellouki, A.; Le Bras, G. J. Phys. Chem. A **2003**, 107, 1499–1505.

(2) Porter, E.; Wenger, J.; Treacy, J.; Sidebottom, H.; Mellouki, A.; Téton, S.; Le Bras, G. J. Phys. Chem. A **1997**, 101, 5770–5775.

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